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(54) Trilayer shutdown battery separator and process of manufacture

Dreilagiger, mikroporöser Batterieseparator und Herstellungsverfahren Séparateur microporeux à trois couches pour cellule électrochimique et procédé de manufacture de celui-ci

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50667 Köln (DE)

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EP-A- 0 872 900 US-A- 4 698 372 GB-A- 2 298 817

 PATENT ABSTRACTS OF JAPAN vol. 013, no. 106 (E-726), 14 March 1989 & JP 63 276868 A (TOKUYAMA SODA CO LTD), 15 November 1988

 PATENT ABSTRACTS OF JAPAN vol. 097, no. 001, 31 January 1997 & JP 08 250097 A (KUREHA CHEM IND CO LTD), 27 September 1996

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Description

Field of the Invention

[0001] This invention generally relates to a trilayer shutdown battery separator, and the method of making same.

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Background of the Invention

[0002] A battery normally comprises electrodes, electrolyte, and battery separators. Battery separators are located between adjacent anodes and cathodes in a battery to prevent direct contact between electrodes of opposite polarity and to contain electrolyte.

[0003] In lithium batteries (e.g., lithium ion or lithium secondary battery) which are increasingly popular, short-circuiting is a problem as it tends to cause thermal runaway and even explosion. Thus, shutdown separators have been developed to cope with this problem. (See, e.g., US Patent No. 4,650,730 and US Patent No. 4,731,304 both issued to Lundquist et al.) A shutdown battery separator is a microporous membrane which closes its pores at some temperature well below the melting and/or ignition point of lithium to minimize the negative impact of thermal runaway.

[0004] Microporous membranes made of polymers such as polypropylene which give rise to higher puncture strength often have a high melting temperature, near the melting point of lithium, which is a drawback for using these polymers to form shutdown separators for lithium batteries. On the other hand, microporous membranes made of polymers such as polyethylene generally have low melting temperature. But their puncture strength is generally low. Thus, trilayer shutdown batteries have been proposed comprising two microporous polypropylene membranes sandwiching a polyethylene membrane.

[0005] Japanese Patent Application No. 98395/1994 (JP7-304110A) filed May 12, 1994 by Ube Industires, Ltd. discloses a microporous trilayer battery separator having a polypropylene-polyethylene-polypropylene construction. The trilayer battery separator has a poreclosing shutdown temperature of 135 to 140°C. The process of preparing the separator includes the steps of: extruding a polypropylene non-porous precursor; extruding a polyethylene non-porous precursor; forming the polypropylene-polyethylene-polypropylene non-porous trilayer precursor; bonding the trilayer precursor together; heat-treating (annealing) the bonded precursor; and stretching the bonded, annealed precursor to form the porous, trilayer separator.

[0006] U.S. Patent No. 5,691,077 to Yu also teaches a trilayer battery separator having a polypropylene-polyethylene-polypropylene construction. The shutdown temperature given in the example is 132°C. The process for making the trilayer battery separator includes the steps of: forming a porous polypropylene precursor;

forming a porous polyethylene precursor; forming a porous trilayer precursor; and bonding the porous trilayer precursor to form the trilayer battery separator.

[0007] UK Patent Publication No. GB 2,298,817 describes a microporous trilayer battery separator made by co-extruding a trilayer film precursor having a nonporous polypropylene-polyethylene-polypropylene construction, annealing the trilayer precursor, and then stretching the annealed trilayer precursor to form the porous trilayer battery separator. The separator has a shutdown temperature of 135°C as given in the examples. [0008] A porous trilayer separator is also proposed in Japanese Patent Application No. 56320/1995 (JP8-250097A) filed by Kureha Chemical Industry Co. Ltd., and is purported to have a shutdown temperature in the range of from 100°C to 150°C. However, in the working examples, the shutdown temperature of the trilayer separators is 124°C. The Kureha separator is prepared by a process that includes the steps of: co-extruding a trilayer precursor, which contains, e.g., a solvent extractable material as pore forming aid, and forming the pores in the precursor by solvent extraction of the extractable-laden precursor.

[0009] A microporous shutdown separator should be thin enough to minimize the space it occupies in the battery and to reduce electrolytic resistance. Nevertheless the shutdown separator must also have sufficient strength to resist splitting and puncture. Although these two attributes, i.e., thinness and strength, are each very important, the two cannot be maximized since film strength typically varies inversely with film thickness. Moreover, it is also desirable to provide a separator having a shutdown temperature lower than 120°C, preferably within the range of from 95°C to 115°C, while having a higher temperature at which the integrity of the separator can be maintained. As discussed above, the previously disclosed trilayer shutdown separators all have a shutdown temperature higher than 120°C. This is largely because the methods heretofore known for lowering shutdown temperature either compromise separator thinness or substantially weaken separator strength and interfere with the ability to manufacture the separator. Consequently, there has not been available in the art a trilayer separator having a shutdown temperature of lower than 120°C while still possessing satisfactory thinness as well as sufficient strength. Thus, there is further need for high quality battery separators.

Summary of the Invention

[0010] The three desirable features, i.e., satisfactory thinness, sufficient strength, and relatively low shutdown temperature, are achieved in the trilayer shutdown battery separator provided in the present invention. The trilayer shutdown battery separator has two microporous strength layers sandwiching one inner microporous shutdown layer. The microporous inner layer is formed by a particle stretch method. Preferably, the shutdown

temperature of the trilayer separator is within the range of from 80°C to 120°C, more preferably from 95°C to 115°C. Methods of making the trilayer shutdown separator are also provided. A preferred method comprises the following steps: (a) extruding non-porous strength layer precursors; (b) annealing and stretching the non-porous precursor to form microporous strength layers; (c) forming a microporous inner layer by a particle stretch method; and (d) bonding two strength layers and one shutdown layer into the trilayer battery separator wherein the first and third layers are strength layers, and the second layer is said microporous membrane made by a particle stretch method.

Description of the Figures

[0011] Figure 1 is a schematic diagram illustrating the three-layer construction of the trilayer battery separator of the present invention.

Detailed Description of the Invention

[0012] The present invention relates to a trilayer shutdown battery separator comprising two strength layers sandwiching one shutdown layer. The shutdown inner layer is formed by a particle stretch method as discussed in detail below.

[0013] Figure 1 illustrates a preferred embodiment of shutdown battery separator (10) which comprises a trilayer microporous film for use in electrochemical cells, e.g., batteries, particularly rechargeable batteries, such as lithium batteries. This trilayer separator has three microporous membranes bonded together. The first (12) and third (16) layers, i.e., the two outer layers, are strength layers. The second (14) layer, i.e., the inner layer, is a shutdown layer. The shutdown layer is capable of melting and closing the pores at a temperature (shutdown temperature) below the melting point of the strength layers and substantially below the temperature at which thermal runaway can occur. Preferably, the shutdown temperature of the trilayer battery separator has a shutdown temperature of less than 124°C, and is in the range of from 80°C to 124°C, more preferably from 95°C to 115°C. Preferably the trilayer battery separator has a thickness of less than 50.8 µm (2 mils), more preferably less than 38.10 µm (1.5 mils), and most preferably no greater than 25.4 µm (1 mil).

[0014] The trilayer battery separator exhibits reduced splitness and good puncture strength while still having satisfactory thinness and desirable low shutdown temperature.

[0015] In addition, as described below, the inner layer contains fillers which generally have much greater heat conductivity than polymer matrix. Therefore, compared with the trilayer battery separators heretofore known in the art, the trilayer separator of the present invention has a more uniform heat distribution during battery operation, and thus promotes a safer battery by preventing

overheating in battery cells.

[0016] The strength layers can be composed of, for example, a polyolefin such as polypropylene or polyethylene, or a blend comprising substantially polypropylene or polyethylene or copolymer thereof. Preferably, polypropylene or a blend formed substantially entirely of polypropylene (e.g., 95% by weight of polypropylene or greater) is used as the polymer forming the strength layers. An exemplary polypropylene is Fina PP 3271 resin commercially available from Fina Oil and Chemical Company, Dallas Texas.

[0017] The strength layers can be made with methods known in the art for making microporous membranes, including but not limited to a stretch method, an extraction method (see, e.g. Japanese Patent Application No. 56320/1995 (JP 8250097A), filed by Kureha Chemical Industry Co. Ltd. on March 15, 1995), and a particle stretch method (see, e.g., U.S. Patent Nos. 3,870,593 and 4,777,073).

20 [0018] Preferably, they are made by a stretch method which involves subjecting pre-formed non-porous membranes to uniaxial or biaxial stretching to make them microporous.

[0019] Broadly speaking, the preferred stretch method for making the outer strength layers comprises extruding, e.g., a polypropylene polymer or copolymer to form a membrane, annealing the membrane, and stretching the annealed membrane. By way of nonlimiting example, suitable methods for this purpose are disclosed in U.S. Patent Nos. 3,426,754; 3,588,764; 3,679,538; 3,801,404; 3,801,692; 3,843,761; 3,853,601; 4,138,459; 4,539,256; 4,726,989; 4,994,335; and 5,565,281.

[0020] In particular, the improved methods for making thinner films disclosed in EP 0 718 901 A and EP 0 892 448 (U.S. Patent application Serial No. 08/650,210 filed May 20, 1996), and U.S. Patent 6,057,060 (Application Serial No. 08/896,513 filed June 22, 1997), are preferred.

[0021] The inner shutdown layer of the present invention is made by a particle stretch method. By "particle stretch method" is intended a method of forming a microporous film comprising stretching a precursor film made from a polymer matrix filled with solid polymer fillers dispersed therein. The stretching results in pore formation due to stress concentration, whereby the film is rendered microporous. Any particle stretch method known in the art may be used for making the inner layer of this invention. Examples of such methods can be found in, for example, U.S. Patent Nos. 3,870,593; 4,350,655; 4,698,372; and 4,777,073.

[0022] Specifically, the shutdown layer of this invention is produced by stretching a precursor film that is extruded from a polymer composition comprising a polymer and a filler. Optionally, the polymer composition may also include conventional additives such as stabilizers and antioxidants.

[0023] Any polymer suitable for film production may

be used for making the inner layer of the trilayer battery separator of this invention. Examples of such polymer include, but not limited to, polyolefins, polysulfones, polyvinyl choride, polyvinyl fluoride, polytetrafluoroethylene-polystyrene copolymers, polyamides, polyphenyleneoxide-polysterene copolymers and polycarbonates. Preferably, polyolefins are used. Exemplary polyolefins include but not limited to polypropylene, polyethylene, and blends comprising substantially one of the above compounds.

[0024] Preferably, polymers should be chosen so that the shutdown temperature falls within the range of from 80°C to 120°C, preferably from 95°C to 115°C. To achieve lower shutdown temperatures, it is preferred to use low density polyethylene (LDPE), linear low density polyethylene (LLDPE), or a blend comprising substantially LDPE, LLDPE or a mixture thereof. Particularly preferred is LLDPE, a random copolymer of ethylene and alpha-olefins selected from the group consisting of C₃₋₁₀ alpha-olefins having a polymerized alpha-olefin content of 20 (preferably 16)% by weight. The preferred examples of LLDPE include but not limited to copolymers of ethylene-butene and copolymers of ethylene-butene

[0025] Fillers useful in this invention must have a low affinity for and a significantly lower elasticity than polymer component. At the same time, fillers should have some interaction with the polymer. The fillers used must also have a melting point higher than that of the polymer used in the polymer composition. In addition, suitable filler preferably are non-hygroscopic and water-insoluble. By "non-hygroscopic" is intended that the filler does not absorb moisture from the air. Fillers may be any inorganic or organic material. Preferably the filler should be a rigid material having a non-smooth surface.

[0026] Examples of suitable fillers include but not limited to alumina, mica, barium carbonate, barium sulfate, calcium carbonate, calcium oxide, calcium sulfate, clay, diatomaceous earth, glass powder, kaolin, magnesium carbonate, magnesium sulfate, magnesium oxide, silica, silica clay, talc, titanium oxide and zinc oxide, Examples of organic fillers include but not limited to polymer powders such as powders of poly(hexamethylene adipamide), poly(ethylene terephthalate) and beads of poly (styrene divinyl benzene).

[0027] Calcium carbonate is particularly preferred for use as the filler in this invention. In addition, the inorganic fillers such as calcium carbonate are preferably surface treated by coating on their surface with, for example, calcium stearate.

[0028] The amount of filler added to the polymer depends on the desired properties of the microporous film, including mechanical performance, permeability, and electrical resistance. Generally, the amount of filler used varies with different fillers. For example, in a polymer composition comprising LLDPE as polymer and calcium carbonate as filler, the amount of calcium carbonate preferably is from 35 to 65 percent by weight of the polymer.

ymer composition. Minor experiment may be required to determine the optimum amount of filler with respect to different polymers and fillers, this being well within the capability of one skilled in the art once apprised of the present disclosure.

[0029] The particle size of the filler is determined to be very critical in this invention to make low temperature shutdown inner layer having a lower electric resistance. Preferably, smaller particles should be used in order to produce thinner microporous film. Furthermore, it is discovered for the first time that smaller filler particles leads to desirable lower electric resistance. Among the commercially available fillers, it is found that a particle size of from 0.1 to 1 μ m (micron) is most appropriate for producing thin microporous films having low electric resistance.

[0030] The polymer(s) and filler(s) in the polymer composition for making the precursor film for the inner layer must be uniformly and intimately mixed. Any suitable method of mixing known in the art can be used. For example, mixing may be achieved by premixing the polymer and filler together and then passing the blend through a compounding extruder. Mixing may also be conducted by feeding the polymer and filler components directly to a mixing device. Suitable mixing device include but not limited to a compounding extruder, high shear continuous mixer, two roll mill, or an internal mixer such as a Banbury mixer.

[0031] In accordance with the present invention, a precursor film can be extruded from the intimately mixed polymer composition described above, and the precursor film is then made microporous by stretching.

[0032] Extrusion can be achieved by methods generally known in the art including but not limited to cast extrusion and blown film extrusion.

[0033] Stretching or "orientation" of the precursor film may be carried out in conventional methods. The precursor film can be stretched uniaxially, either in the machine direction or in the transverse direction. It can also be stretched biaxially in both the machine direction and the transverse direction. When the film is biaxially stretched, the stretching in the two directions can be done either simultaneously or sequentially. When the polymer used in the polymer composition is LLDPE, blown films are preferably stretched in the machine direction or in both directions whereas cast films are preferably stretched in the transverse direction. Stretching can be done in conventional methods generally known in the art. An example of a preferred method is the one that is disclosed in U.S. Patent No. 4,777,073 to Sheth. [0034] After the microporous outer layers and microporous inner layer are made, they are laminated and bonded together into a trilayer battery separator having a structure as shown in Figure 1. The bonding methods are generally known in the art. Suitable methods for use in this invention include calendaring, adhering with adhesives, and welding. The application of adhesives may include: air atomizing; gravure/screen printing; hydrau-

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lic spraying; and ultrasonic spraying. The choice of adhesive and the rate of adhesive application must be chosen so that the separator's porosity is not adversely effected. The welding technique includes thermal welding and ultrasonic welding. The amount of energy for either welding procedure and the pattern of weld should be chosen so that the separator's porosity is not adversely effected. Preferably bonding is done by calendaring, with nips closed, at a temperature of at least 1°C below the melting point of the inner shutdown layer, preferably at a temperature of at least 5°C below the melting point of the inner shutdown layer.

[0035] Optionally, a cross-plied trilayer battery separator can be made by cross-ply laminating the microporous layers such that one strength layer is angularly oriented relative to the other strength layer. Suitable methods of making cross-ply microporous battery separator are disclosed in U.S. Patent No. 5,667,911.

[0036] After bonding, the trilayer shutdown battery separator is rewound for use in the manufacture of batteries, particularly secondary lithium batteries, as is well known in the art.

[0037] The invention is further illustrated with reference to the examples set forth below.

Examples

EXAMPLE 1

[0038] A linear low density polyethylene(LLDPE) with a melt' flow index of 2.0 was compounded with calcium carbonate particles that are surface-treated with calcium stearate. The calcium carbonate has an average particle size of 1 µm (micron). Polymer compositions having LLPDE filled with 40%, 45%, and 50% by weight of CaCO3 respectively were then cast extruded into precursor films, each having a thickness of approximately 50.8 to 203.20 µm (2 to 8 mil). Each resulting precursor film was cooled and subjected to tenter stress in the transverse direction with a stretch ratio of 2.5 to 1. Stretching was performed in a tentering frame at approximately 65EC. One layer of the resulting white porous film was then sandwiched between two layers of Celgard® stretched polypropylene microporous membranes and bonded to form a trilayer battery separator. [0039] This separator was tested to determine the electrical impedance versus temperature behavior. The testing method used herewith is described by R. Spotnitz et al. in Proceedings from the 12th International Seminar on Primary and Secondary Battery Technology and Application, March 6, 1995. The impedance was measured as the temperature was raised at a rate of 60°C/min. The impedance analysis showed an increase of impedance near 115°C corresponding to melting of the shutdown inner layer, and a further increase in impedance occurs near 160°C due to melting of the polypropylene outer layers.

[0040] Although the foregoing invention has been de-

scribed in some detail by way of illustration and example for purposes of clarity of understanding, it will be obvious that certain changes and modifications may be practiced within the scope of the appended claims.

Claims

- A battery separator comprising a first and third microporous strength layer sandwiching a second shutdown layer, said shutdown layer being a microporous membrane made by a particle stretch process.
- A battery separator of Claim 1, wherein said battery separator has a shutdown temperature of about 115°C.
 - A battery separator of Claim 1, wherein said shutdown layer comprises calcium carbonate particles as fillers.
 - A battery separator of Claim 1, wherein said shutdown layer comprises a filler having a particle size of from 0.1 μm (0.1 micron) to 1 μm (. micron).
 - **5.** A battery separator of Claim 1, wherein said strength layers are made of polypropylene and said shutdown layer is made of polyethylene.
 - A battery separator of Claim 5, wherein said polyethylene is linear low density polyethylene.
 - A battery separator of Claim 1, wherein said strength layer is made by a stretch method.
 - A battery separator according to claim 1 comprising:
 - a first and third microporous polypropylene layer having a thickness of no greater than 12.7 μm (0.5 mil), each of said polypropylene layer being made by a method comprising the steps of: extruding a parison;
 - collapsing the parison onto itself to form a flat sheet comprising two plies, annealing the flat sheet:
 - stretching the flat sheet, wherein the stretching comprising stretching in at least two steps and each at a different temperature; and winding up the flat sheet, an adhesion force between the two plies being less than 3.15 g/cm (8 grams per inch);
 - a second microporous shutdown layer being made by a particle stretch process and comprising linear low density polyethylene and a calcium carbonate particle filler having a particle size of from about 0.1 μm (micron) to 1 μm

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(micron), said shutdown layer being sandwiched between said first and third microporous polypropylene layer.

A method of making a trilayer battery separator according to claim 1 comprising:

> forming a microporous shutdown layer membrane by a particle stretch method;

> forming microporous strength layer membranes;

> bonding said membranes into the trilayer battery separator wherein the first and third membranes are strength layers, and the second membrane is said shutdown layer formed by a particle stretch method.

- 10. A method of making a trilayer battery separator of Claim 9, wherein said battery separator has a shutdown temperature of 115°C.
- 11. A method of making a trilayer battery separator of Claim 9, wherein said shutdown layer comprises calcium carbonate particles as filler.
- 12. A method of making a trilayer battery separator of Claim 9, wherein said shutdown layer comprises a filler having a particle size of from 0.1 μm (0.1 micron) to 1 μm (micron).
- 13. A method of making a trilayer battery separator of Claim 9, wherein said strength layers are made of polypropylene and said shutdown layer is made of polyethylene.
- 14. A method of making a trilayer battery separator of Claim 13, wherein said polyethylene is linear low density polyethylene.
- 15. A method of making a trilayer battery separator according to Claim 9, wherein said microporous strength layers are made by a stretch method.
- 16. A method of making a trilayer battery separator according to claim 1 comprising:

making a first and third microporous polypropylene layer having a thickness of no greater than 12.7 μ m (0.5 mil) comprising for each of the besaid layers the steps of: extruding a parison; collapsing the parison onto itself to form a flat sheet comprising two plies; annealing the flat sheet; stretching the flat sheet, wherein the stretching comprising stretching in at least two steps and each at a different temperature; and winding up the flat sheet, an adhesion force between the two plies being less than 3.15 g/cm (8 grams per inch);

making a second microporous shutdown layer by a particle stretch process which comprises the steps of: providing a blend comprising of a linear low density polyethylene and calcium carbonate particle fillers having a particle size of from 0.1 μ m (0.1 micron) to 1 μ m (1 micron); extruding a precursor membrane from the blend; and stretching said precursor membrane to impart micropores; and bonding said polypropylene layers and said shutdown layer into the trilayer battery separa-

bonding said polypropylene layers and said shutdown layer into the trilayer battery separator wherein the first and third membranes are the polypropylene layers, and the second membrane is said shutdown layer.

Patentansprüche

- Batterieseparator, umfassend eine erste und eine dritte mikroporöse Festigkeitsschicht, zwischen denen eine zweite Abschaltschicht angeordnet ist, wobei die Abschaltschicht eine mikroporöse Membran ist, die mittels eines Teilchenstreckverfahrens hergestellt ist.
- Batterieseparator nach Anspruch 1, wobei der Batterieseparator eine Abschalttemperatur von etwa 115 °C aufweist.
- Batterieseparator nach Anspruch 1, wobei die Abschaltschicht Calciumcarbonatteilchen als Füllmittel umfasst.
- Batterieseparator nach Anspruch 1, wobei die Abschaltschicht ein Füllmittel mit einer Teilchengröße von 0,1 μm bis 1 μm umfasst.
 - Batterieseparator nach Anspruch 1, wobei die Festigkeitsschichten aus Polypropylen bestehen und die Abschaltschicht aus Polyethylen besteht.
 - Batterieseparator nach Anspruch 5, wobei das Polyethylen lineares Polyethylen niedriger Dichte ist.
- Batterieseparator nach Anspruch 1, wobei die Festigkeitsschicht mittels eines Streckverfahrens hergestellt wird.
 - 8. Batterieseparator nach Anspruch 1, umfassend: eine erste und eine dritte mikroporöse Polypropylenschicht mit einer Dicke von nicht mehr als 12,7 μm (0,5 mil), wobei jede der Polypropylenschichten mittels eines Verfahrens hergestellt ist, umfassend die Schritte des: Extrudierens eines Rohlings; Zusammenfallenlassens des Rohlings auf sich selbst unter Bildung einer flachen, zwei Lagen umfassenden Folie; Spannungsfreimachens der flachen Folie; Streckens der flachen Folie, wobei das Strecken

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das Strecken in wenigstens zwei Stufen und jeweils bei einer verschiedenen Temperatur umfasst, und des Aufwickelns der flachen Folie, wobei die Adhäsionskraft zwischen den beiden Lagen niedriger als 3,15 g/cm (8 g/inch) ist, wobei eine zweite mikroporöse Abschaltschicht mittels eines Teilchenstreckverfahrens hergestellt wird und lineares Polyethylen mit niedriger Dichte und einen Calciumcarbonatfüllstoff mit einer Teilchengröße von etwa 0,1 µm bis 1 µm umfasst, wobei die Abschaltschicht zwischen der ersten und der dritten mikroporösen Polypropylenschicht angeordnet ist.

- 9. Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 1, umfassend: die Bildung einer mikroporösen Abschaltschichtmembran mittels eines Teilchenstreckverfahrens; die Bildung von mikroporösen Festigkeitsschichtmembranen; das Verbinden der Membranen unter Bildung des dreischichtigen Batterieseparators, wobei die erste und die dritte Membran Festigkeitsschichten sind und die zweite Membran die mittels eines Teilchen-Streckverfahrens gebildete Abschaltschicht ist.
- Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 9, wobei der Batterieseparator eine Abschalttemperatur von 115 °C aufweist.
- Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 9, wobei die Abschaltschicht Calciumcarbonatteilchen als Füllmittel umfasst.
- 12. Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 9, wobei die Abschaltschicht ein Füllmittel mit einer Teilchengröße von 0,1 μm bis 1 μm umfasst.
- 13. Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 9, wobei die Festigkeitsschichten aus Polypropylen bestehen und die Abschaltschicht aus Polyethylen besteht.
- Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 13, wobei das Polyethylen lineares Polyethylen niedriger Dichte ist.
- 15. Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 9, wobei die mikroporösen Festigkeitsschichten mittels eines Streckverfahrens hergestellt werden.
- 16. Verfahren zur Herstellung eines dreischichtigen Batterieseparators nach Anspruch 1, umfassend:

die Herstellung einer ersten und einer dritten mikroporösen Polypropylenschicht mit einer Dicke von nicht mehr als 12,7 μm (0,5 mil), umfassend für jede der Schichten die Schritte des: Extrudierens eines Rohlings; Zusammenfallenlassens des Rohlings auf sich selbst unter Bildung einer flachen, zwei Lagen umfassenden Folie; Spannungsfreimachens der flachen Folie; Streckens der flachen Folie, wobei das Strecken das Strecken in wenigstens zwei Stufen und jeweils bei einer verschiedenen Temperatur umfasst, und des Aufwickelns der flachen Folie, wobei die Adhäsionskraft zwischen den beiden Lagen niedriger als 3,15 g/cm (8 g/inch) ist;

Herstellens einer zweiten mikroporösen Abschaltschicht durch ein Teilchenstreckverfahren, umfassend die Schritte des: Bereitstellens eines Blends, umfassend lineares Polyethylen niedriger Dichte und Calciumcarbonatteilchen-Füllmittel mit einer Teilchengröße von 0,1 µm bis 1 µm; Extrudierens einer Vorstufenmembran aus dem Blend und des Streckens der Vorstufenmembran zur Erzeugung von Mikroporen, und des

Verbindens der Polypropylenschichten und der Abschaltschicht unter Bildung des dreischichtigen Batterieseparators, wobei die erste und die zweite Membran Polypropylenschichten sind und die zweite Membran die Abschaltschicht ist

Revendications

- Séparateur de batterie comprenant des première et troisième couches résistantes microporeuses entourant une deuxième couche servant de couche d'arrêt, ladite couche d'arrêt étant une membrane microporeuse produite par un procédé d'étirage en présence de particules.
- Séparateur de batterie suivant la revendication 1, qui a une température d'arrêt d'environ 115°C.
 - Séparateur de batterie suivant la revendication 1, dans lequel la couche d'arrêt comprend des particules de carbonate de calcium comme charge.
 - Séparateur de batterie suivant la revendication 1, dans lequel la couche d'arrêt comprend une charge ayant un diamètre de particules de 0,1 μm (0,1 micromètre) à 1 μm (1 micromètre).
 - Séparateur de batterie suivant la revendication 1, dans lequel les couches résistantes sont consti-

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tuées de polypropylène et la couche d'arrêt est constituée de polyéthylène.

- 6. Séparateur de batterie suivant la revendication 5. dans lequel le polyéthylène consiste en polyéthylène linéaire basse densité.
- 7. Séparateur de batterie suivant la revendication 1, dans lequel la couche résistante est produite par un procédé d'étirage.
- 8. Séparateur de batterie suivant la revendication 1, comprenant:

des première et troisième couches de polypropylène microporeuses ayant une épaisseur non supérieure à 12,7 µm (0,5 mil), chacune desdites couches de polypropylène étant produite par un procédé comprenant les étapes consistant : à extruder une paraison ; à provoquer l'affaissement de la paraison sur elle-même pour former une feuille plate comprenant deux nappes ; à soumettre à un recuit la feuille plate;

à étirer la feuille plate, l'étirage comprenant un étirage en au moins deux étapes, chacune de ces étapes étant mise en oeuvre à une température différente ; et à enrouler la feuille plate, la force d'adhérence entre les deux nappes étant inférieure à 3,15 g/cm (8 grammes par inch):

une deuxième couche consistant en une couche d'arrêt microporeuse produite par un procédé d'étirage en présence de particules et comprenant du polyéthylène linéaire basse densité et une charge constituée de particules de carbonate de calcium ayant un diamètre de particules d'environ 0,1 µm (1 micromètre) à 1 μm (1 micromètre), ladite couche d'arrêt étant intercalée entre lesdites première et troisième 40 couches de polypropylène microporeuses.

- 9. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 1, comprenant les étapes consistant :
 - à former une membrane servant de couche d'arrêt microporeuse par un procédé d'étirage en présence de particules ;
 - à former des membranes servant de couches résistantes microporeuses;
 - à lier lesdites membranes dans le séparateur de batterie en trois couches, les première et troisième membranes étant des couches résistantes et la deuxième membrane étant la couche d'arrêt formée par un procédé d'étirage en présence de particules.

- 10. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 9. dans lequel le séparateur de batterie a une température d'arrêt de 115°C.
- 11. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 9, dans lequel la couche d'arrêt comprend des particules de carbonate de calcium comme charge.
- 12. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 9, dans lequel la couche d'arrêt comprend une charge ayant un diamètre de particules de 0,1 µm (0,1 micromètre) à 1 µm (1 micromètre).
- 13. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 9, dans lequel les couches résistantes sont constituées de polypropylène et la couche d'arrêt est constituée de polyéthylène.
- 14. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 13, dans lequel le polyéthylène consiste en polyéthylène linéaire basse densité.
- 15. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 9. dans lequel les couches résistantes microporeuses sont produites par un procédé d'étirage.
- 16. Procédé pour la production d'un séparateur de batterie en trois couches suivant la revendication 1, comprenant les étapes consistant :

à produire des première et troisième couches de polypropylène microporeuses avant une épaisseur non supérieure à 12,7 μm (0,5 mil), comprenant, pour chacune desdites couches, les étapes consistant : à extruder une paraison ; à provoquer l'affaissement de la paraison sur elle-même pour former une feuille plate comprenant deux nappes ; à soumettre à un recuit la feuille plate; à étirer la feuille plate, l'étirage comprenant un étirage en au moins deux étapes, chacune à une température différente ; et à enrouler la feuille plate, la force d'adhérence entre les deux nappes étant inférieure à 3,15 g/cm (8 g/inch);

à produire une deuxième couche consistant en une couche d'arrêt microporeuse par un procédé d'étirage en présence de particules qui comprend les étapes consistant : à prendre un mélange constitué d'un polyéthylène linéaire basse densité et d'une charge consistant en particules de carbonate de calcium ayant un diamètre de particules de 0,1 µm (0,1 micromètre) à

1 μm (1 micromètre) ; à extruder une membrane servant de précurseur à partir du mélange ; et à étirer ladite membrane servant de précurseur pour former des micropores; et à lier lesdites couches de polypropylène et la- 5 dite couche d'arrêt dans le séparateur de batterie en trois couches dans lequel les première et troisième membranes consistent en les couches de polypropylène et la deuxième membrane consiste en ladite couche d'arrêt.

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FIG. 1

